STUDY ON FRACTURE PROPAGATION OF HYDRAULIC AND SUPERCRITICAL CO₂ FRACTURING IN DIFFERENT ROCK

by

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In this study, the performance of water-based fracturing fluids and supercritical CO₂ in three types of representative reservoirs (sandstone, conglomerate, and shale) was investigated. The results showed that there are differences in fracture initiation pressures in different rocks, but the fracture initiation pressure of supercritical CO₂ is lower than that of water regardless of the rock type. In sandstone reservoirs, supercritical CO₂ induced more complex fractures than water, resulting in branching fractures. In conglomerate reservoirs, hydraulic fractures pass through the conglomerate and are flatter, whereas supercritical CO₂ fractures pass through or around the conglomerate, and thus the fractures are more tortuous. Gravel stopped the fracture extension in both conditions. In shale reservoirs, supercritical CO₂ can communicate natural fractures more effectively than water, thereby increasing the effective transformation volume. The study provides theoretical guidance for reservoir adaptation of supercritical CO₂ fracturing.

Keywords: Hydraulic fracturing, Supercritical CO₂, Initiation pressure, Fracture pattern

Introduction

Unconventional oil and gas reservoirs have dense rocks with extremely low porosity and permeability, requiring hydraulic fracturing to construct oil and gas seepage pathways for commercial

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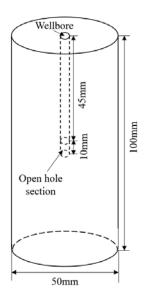
development. Water-based fracturing fluids are widely used to crush the reservoir and carry the proppant to form highly conductive fractures. However, there are some shortcomings of water-based fracturing fluids in the application process. Water swells clay minerals and softens the fracture surface, which embeds the proppant and reduces the hydraulic conductivity [1-3]. The water required for hydraulic fracturing wells is measured in tens of thousands of cubic meters, which consumes a large amount of water resources, while the return fluid will pollute the environment.

Supercritical CO₂ attracts attention with its unique physical and chemical properties. It has an extremely low viscosity and a density close to that of water. Supercritical CO2 replacing water as a fracturing fluid will avoid clay expansion, reduce fracture initiation pressure, and create a complex fracture network in the reservoir [4-6]. Injected CO₂ increases formation pressure and benefits production. In addition, CO₂ can displace adsorbed methane and improve recovery while allowing for carbon sequestration [7]. Numerous experimental and numerical simulation studies have been carried out on CO₂ flow in wells and fractures, fracture initiation and propagation, and proppant transportation [8-12]. Yang Bing et al.[13] used CT to reconstruct the fracture in three dimensions and clarified that supercritical CO₂ produces more complex fractures under smaller differential stress. Zhou Dawei et al.[14] investigated the temperature effect caused by injected CO₂, and the results showed that the reduction of effective stress caused by thermal tensile stress and pore pressure is the main reason for the reduction of fracture pressure and the formation of fracture network in supercritical CO₂ fracturing and that the increase of temperature difference between reservoir and injected CO2 is favorable for fracturing. Wang Haizhu et al. [15] established a wellbore flow model for supercritical CO₂ fracturing return flow and found that reducing the discharge volume and increasing the tubing size can effectively keep the wellbore temperature high and reduce the risk of CO₂ hydrate formation. Zheng Yong et al. [16] investigated the effect of pump injection schedule on proppant transportation in supercritical CO₂ fracturing by CFD-DEM method and found that injecting high-density proppant first followed by low-density proppant and lowering the initial temperature canimprove the performance of supercritical CO₂ carrying proppant.

Nevertheless, the formation adaptability of CO_2 fracturing technology is still unclear. To address this issue, water, and CO_2 fracturing experiments were carried out on three typical reservoir rocks (sandstone, limestone, and conglomerate) in this paper. Characteristics of injection pressure variations were analyzed. The differences between the fracture morphology of water and CO_2 fracturing were compared. The results of the study are a guide to understanding the effectiveness of CO_2 fracturing in reservoirs with different lithologies.

Materials

To compare the differences between water and CO_2 fracturing in reservoirs with different lithologies, three rocks were selected for this paper. Sandstone and limestone were drilled from natural out crops extracted from the Ordos Basin. The conglomerate outcrops collapsed after weathering, making it difficult to drill the sample. Therefore, artificial conglomerates cemented with concrete and gravel were taken. A hole with a depth of 55mm was drilled in the center of a ϕ 50mm*100mm cylindrical sample as a wellbore to inject the fracturing fluid. At the same time, the bottom of the borehole is reserved for a 10mm open hole section, as shown in Figure.1.The experimental scheme is shown in Table 1.



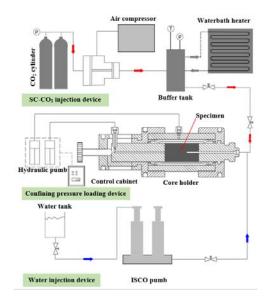


Figure. 1 Schematic diagram of open hole completion

Figure. 2 Schematic diagram of the fracturing experimental setup [10]

Table	1	experimental	design
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No.	Rock type	Fluid type	Temperature/°C	$\sigma_{_{\scriptscriptstyle{V}}}*\sigma_{_{\scriptscriptstyle{r}}}$ /MPa	InjectionRate/mL·min ⁻¹
#1	Sandstone	Water			
#2	Sandstone	CO_2			
#3	Limestone	Water	40	5*15	40
#4	Limestone	CO_2			
#5	Conglomorata	Water			
#6	Conglomerate	CO_2			

Apparatus

The fracturing device used is shown in Figure. 2, which can carry out fracturing experiments with a variety of fluids, including water, carbon dioxide, and nitrogen.

The supercritical CO_2 injection device was used by a pneumatic booster pump to pressurize the low-pressure CO_2 and store it in a high-pressure tank. CO_2 is heated by a water bath to bring it to a supercritical state. A maximum pressure of 100MPa is available. The water injection unit provides water injection at a constant rate or pressure by ISCO pumps. The core holder of the confining pressure loading device can contain samples of ϕ 50mm*100mm, and axial and radial stresses of up to 80MPa could be applied through the hydraulic pump.

Fracturing evolution characteristics

Figure 3 demonstrates the pressure-time profiles of water- and CO₂-induced fractures in different rocks. The pressure reached the peak value and then decreased rapidly. The trends of the curves indicated that all three rocks showed strong brittleness during the fluid-driven damage. The pressure profiles have similar characteristics regardless of the rock type. The pressure profile of CO₂ has a smaller slope at the beginning of injection compared to water. The pressure curves have similar characteristics regardless of the rock. The pressure curve of CO₂ has a smaller slope at the early stage of injection compared to water. This is because CO₂ is more strongly compressible than water, and the pressurization process needs to compress the CO₂ first, so the initial pressurization rate is slower. As a result, CO₂-induced rocks take longer to fracture than water. This also allowed sufficient time for CO₂ to spread through the pores of the rock, increasing the pore pressure and reducing the effective stress.

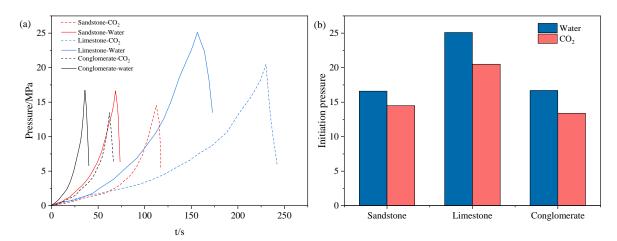


Figure. 3 Injection pressure profile (a) and initiation pressure (b)

 CO_2 fracturing has a lower initiation pressure than water in all three rocks. During the fracturing process, the surface tension of the fluid is directly related to whether it can enter the crack tip[17-18]. In the process of fracture propagation, supercritical CO_2 , which has a lower viscosity and almost zero surface tension, has less flow resistance inside the fracture and could reach the fracture tip. As shown in Figure.4, supercritical CO_2 has a smaller fluid lag zone at the crack tip, which reduces the net pressure required for crack propagation [19].

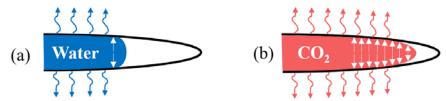


Figure. 4 Schematic diagram of crack tips for different fluids (a) Water and (b) CO₂

Since the rock samples in the experiment were small, crack initiation and extension occurred almost at the same time, and the peak of the pressure curve was assumed to be the initiation pressure, as shown in Figure.3(b). The fracture initiation pressures for CO₂ fracturing in sandstone, chert, and conglomerate were decreased by 2.6 MPa, 4.6 MPa, and 3.1 MPa, respectively. In limestone with high initiation pressure, the effect of CO₂ fracturing in reducing initiation pressure is more significant.

Fracture morphology

The fracture morphology of water and CO_2 fracturing in different rocks is shown in Figure. 5.It can be visualized that hydraulic fracturing produced only one main fracture and CO_2 fracturing induced a more complex fracture.

In sandstones, hydraulic fracturing produces a relatively simple fracture with only one main fracture perpendicular to the wellbore. In contrast, CO_2 -induced fractures have complex geometries. The fractures produced branches and secondary fractures after fracture initiation at the bottom of the wellbore. Due to the presence of laminations in the sandstone, the fracture morphology is also influenced by the laminations. In limestones, hydraulic fracturing produced a main fracture that intersected the wellbore diagonally. The fracture was deflected along the axial stress after initiating from the bottom of the borehole. In addition to the near wellbore, CO_2 fracturing produced a fracture in the lower portion of the rock sample and a partial collapse on the side of the sample.

The presence of conglomerates gives conglomerates a stronger heterogeneity than sandstones and limestones. The strength of the gravel, its content, its size, and cementation strength all affect the

final patterns of the cracks[20-21]. Four modes of fracture-gravel interaction exist: penetration, deflection, attraction, and termination[22]. Penetration and deflection occurred in hydraulic fracturing, and deflection and attraction occurred in CO₂ fracturing in this study. Fractures in conglomerates were significantly more tortuous than in sandstones and limestones. Most of the CO₂-induced fractures in conglomerates are deflections rather than penetrations. It is explained by the fact that CO₂ has a much lower viscosity and can enter the interface between the conglomerate and the matrix. And it is easier to open the interface than to crush the matrix and gravel. Moreover, matrix collapse can be seen near the gravels in the CO₂-fractured rock samples. These disintegrated particles can act as proppants to form self-supporting fractures.

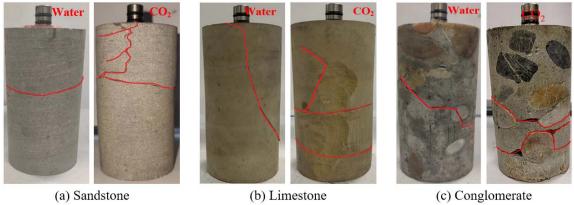


Figure. 5 Fracture morphology of water and CO₂ fracturing in different rocks

The lengths of water- and CO_2 -induced fractures in three rocks were counted, as shown in Fig.6. CO_2 fracturing forms multiple fractures, so the fracture lengths are significantly longer than those of hydraulic fracturing. The lengths of CO_2 -induced fractures in sandstone, limestone, and conglomerate were 2.56, 1.86, and 2.15 times the lengths of those induced by water, respectively.

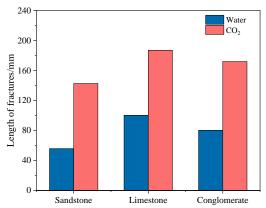


Figure. 6Length of fractures induced by water and CO₂ fracturing in different rocks

Conclusion

Several important conclusions can be drawn from the current experimental studies of hydraulic fracturing performed by injecting water and CO₂into different rocks: CO₂ could reduce the initiation pressure of rock, and this effect is more pronounced in dense and stiff reservoirs. The initiation pressure of CO₂ fracturing in limestone was 4.6 MPa lower than that of water. For different rocks, CO₂ fracturing produces more tortuous fractures with more complex spatial patterns, which provide more efficient flow channels for oil and gas transportation. Enhancement of CO₂ fracturing in conglomerate reservoirs is more pronounced than in sandstones and limestones. In dense, hard, heterogeneous water-

sensitive reservoirs, CO₂ is more preferred than water as a fracturing fluid. The results of the study provide a reference for the selection of fracturing fluids for reservoirs with different lithologies. In reservoirs with a high content of carbonate minerals, the dissolving effect of CO₂ on minerals cannot be neglected, which needs further study.

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