# EXERGETIC PERFORMANCE OF VAPOR-COMPRESSION REFRIGERATION SYSTEM WITH TiO<sub>2</sub> NANOADDITIVE IN THE COMPRESSOR OIL

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Exergy analysis of a vapor-compression refrigeration system with  $TiO_2$  nanoadditives in the compressor oil was performed. Two-step method was used for the preparation of nanooil for various solid particle volume fractions between 0% and 1%. Irreversibilities were determined by using the Second law of thermodynamics. It is found that a reduction in total irreversibility is achieved with nanoparticle inclusion and it was significant for higher particle volume fraction. Key words: refrigeration, nanooil, thermodynamic, irreversibility

### Introduction

Recently, nanosized particles (metallic or non-metallic) are added to heat transfer fluids such as water or ethylene glycol to enhance heat transfer performance in various thermal engineering applications. In refrigeration systems, when nanoadditives are added to the refrigerant or to the compressor oil, it is possible to obtain energy efficient systems [1-10]. Thus, electric energy consumption is reduced and possible detrimental environmental effects are diminished.

In a review study, Celen *et al.* [11] examined the fluid flow characteristics and applications of nanorefrigerants. They noted that with the use of nanorefrigerants, electric energy consumption was reduced and coefficient of performance was enhanced. Mahbubul *et al.* [12], examined the heat transfer and pressure drop of Al<sub>2</sub>O<sub>3</sub>-R134a nanorefrigerant in a smooth pipe with existing correlations. They showed that pressure drop and heat transfer rate was significantly enhanced and there was an optimum value of the nanoparticle volume fraction to obtain best performance. In the experimental study of Bi *et al.* [13], HCC134a refrigerant was used with mineral oil and TiO<sub>2</sub> nanoparticles. For 0.1 wt.% amount of TiO<sub>2</sub>, energy consumption was reduced by 26.1%. They performed the same tests for Al<sub>2</sub>O<sub>3</sub> nanoparticles and showed that nanoparticle type has little effect on the performance of refrigerator. Naphon *et al.* [14] analyzed the effects of nanorefrigerant with 21 nm diameter titanium nanoparticles on the efficiency of a heat pie. They obtained 40% more efficient system for nanorefrigerant containing 0.1 wt.% nanoparticles. Peng *et al.* [15], experimentally examined the boiling heat transfer with R113 refrigerant and CuO nanoparticles. They made the experiments for nanoparticle concentration of 0% and 0.5%. They obtained correlations for various heat flux,

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mass-flow rates, and amount of vapor under constant evaporation pressure. They obtained maximum enhancement of 29.7% in the average heat transfer rate. In the study by Bi *et al.* [16], R600a-TiO<sub>2</sub> nanorefrigerant performance for a refrigerator was tested. They obtained 9.6% of energy saving with 0.5 g/L of TiO<sub>2</sub>-R600a nanorefrigerant as compared to system using only pure R600a refrigerant. Other studies that may be mentioned are found in [17-19].

Based on the previous literature survey, it is observed that adding nanoparticles to the refrigeration systems results in enhancement of coefficient of performance of these systems. In the literature, there exists a vast amount of energy-exergy analysis of refrigeration systems [20-27], but there is no study was found that performed exergy analysis of system containing nanoadditives. In this study, thermodynamic analysis including exergy analysis for a vapor-compression refrigeration system will be performed for a system containing R134 refrigerant and TiO<sub>2</sub> nanoparticles that were used in the compressor oil.

# Method

A vapor-compression refrigeration system was considered in this study. The R134-a was used as refrigerant and polyalkylene glycol oil was augmented with TiO<sub>2</sub> nanoadditives. The system consists of compressor, condenser, expansion device, and evaporator. The condenser and evaporator are finned air cooled and their fan is driven by an electric motor. A 0.375 kW semi-hermetic type compressor (reciprocating) was used. Measurements were made in



Figure 1. Vapor-compression refrigeration system; elements: condenser, evaporator, expansion valve, pressure and temperature indicators, max-min pressure control

seven locations for temperature and two pressure values were used in the system. A photographic view of experimental system with temperature/pressure indicators are depicted in fig. 1. For each of the experimental trial, test periods of 1 hour were considered. This time was required to have steady flow conditions for each run. For each test (considering one solid particle volume fraction), three runs were performed and average values were recorded. The experimental room temperature was measured and it was assured that the tests are performed under the same operating room temperature conditions.

# Nanooil

The polyalkylene glycol oil was used with refrigerant R134a and it is widely used in automotive refrigeration system. It can be fully mixed with R134a refrigerant and compatible with refrigeration system components and it has long-lasting and stable characteristics. To in-

crease the efficiency of oil, nanosized  $TiO_2$  additives were added. Nanoparticles are spherical shaped and the average particle size is 15 nm. The nanooil was prepared in the R&D lab of Ege Nanotek Company which is located in Izmir, Turkey. Two-step method was utilized and stable, long lasting homogenous mixture were obtained. In the preparation of nanooil two step method was used and no surfactants were included. The stability and the possibility of any agglomeration can be checked by measuring the zeta potential of the nanooil. To have good stability, the zeta potential of the nanooil sample should be high. In the current work, the pro-

ducer of the compressor oil ensured that the mixture T has a higher zeta potential value when the experimental study was performed with the nanoparticle enhanced oil.

Agglomeration of nanoparticles results in heat transfer degradation especially for the condenser and evaporator heat exchangers within the system where effective heat transfer takes place. For long term use, to avoid agglomeration, ultrasonic mixers can be uti-

able 1	. Nanoa	dditive	TiO <sub>2</sub>	properties
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Average particle size	15 nm	
Purity	99.5%	
Surface area	60 m²/g	
Density	3.9 g/cm <sup>3</sup>	
Shape	Spherical	

lized to homogenize the mixture. Properties of TiO<sub>2</sub> nanoparticle is given in tab. 1.

# **Exergy analysis**

For a steady flowing device (open system), energy conservation and entropy generation equations can be written:

$$\dot{Q} - \dot{W} = \sum \dot{m}_{out} \left( h_{out} + \frac{V_{out}^2}{2} + gz_{out} \right) - \sum \dot{m}_{in} \left( h_{in} + \frac{V_{in}^2}{2} + gz_{in} \right)$$
(1)

$$\dot{S}_{gen} = \sum \dot{m}_{out} S_{out} - \sum \dot{m}_{in} S_{in} + \frac{Q_{env}}{T_0}$$
<sup>(2)</sup>

In the present study, kinetic and potential energy terms are neglected. The physical specific exergy expressed:

$$\psi = (h - h_0) - T_0(S - S_0) \tag{3}$$

The irreversibility is obtained by using an exergy balance:

$$\dot{I} = \dot{m}_{\rm in}\psi_{\rm in} - \dot{m}_{\rm out}\psi_{\rm out} + Ex_Q + Ex_{\rm W} \tag{4}$$

where  $Ex_Q$  and  $Ex_W$  denote the exergy flow associated with heat transfer and work. Considering each element of the vapor-compressor refrigeration system separately, energy and exergy balance for each of the components are given in tab. 2.

Table 2. Energy and exergy balance for components of refrigeration system

Component	Energy balance	Exergy balance
Compressor	$\dot{W_{\rm c}} = \dot{m}(h_2 - h_1)$	$\dot{I} = \dot{m}(\psi_1 - \psi_2) + \dot{W_c}$
Condenser	$\dot{Q}_{\rm c} = \dot{m}(h_2 - h_3)$	$\dot{I} = \dot{m}(\psi_2 - \psi_3) - \dot{Q}_{\rm c} \left(1 - \frac{T_0}{T_c}\right)$
Expansion valve	$h_3 = h_4$	$\dot{I} = \dot{m}(\psi_3 - \psi_4)$
Evaporator	$\dot{Q}_{\rm c} = \dot{m}(h_1 - h_4)$	$\dot{I} = \dot{m}(\psi_1 - \psi_4)$

# **Results and discussion**

In this study, exergy analysis of a vapor-compression refrigeration system with TiO<sub>2</sub> nanoadditives in its compressor oil were performed. Irreversibility for various system compo-

nents was obtained by measuring of temperature and pressure as indicated in tab. 2 by using exergy balance. Figure 2 shows the irreversibility for different components in the absence of nanoadditives. The highest irreversibility is obtained for compressor and lowest one for expansion valve. For other components, the irreversibility is similar with that of compressor. When 0.5% TiO<sub>2</sub> was added, significant enhancement and reduction in the irreversibilities for the compressor and condenser components were obtained, fig. 3. For this amount of TiO<sub>2</sub> nanoadditive, the viscosity of nanooil increases which results in irreversibility enhancement. In the expansion valve, there are slight changes in the irreversibility with nanoparticle addition.



Figure 2. Irreversibility per unit mass [kWkg<sup>-1</sup>] in various components in the absence of nanoparticles



Figure 3. Irreversibility per unit mass [kWkg<sup>-1</sup>] in various components for nanoparticle volume fraction of 0.05%

At the highest solid particle volume fraction, irreversibility for the compressor reduces from 19.65 to 12.06, fig. 4. This shows that although the viscosity of the nanooil in-



Figure 4. Irreversibility per unit mass [kWkg<sup>-1</sup>] in various components for nanoparticle volume fraction of 1%



creases for high volume fraction of particles, the potential improvement in the compressor was enhanced. The amount of irreversibility in the system where nanoparticles are used at different volume fractions varies for various system components. The component with the least variation is the expansion valve for which the least the irreversibility takes place.

Table 3 shows the irreversibility per unit mass and reduction in percentage for various nanoparticle volume fractions. Total irreversibility of the system decreases with increasing volume fraction of particle. It is observed that the total irreversibility changes slightly for

 Table 3. Irreversibility per unit mass for various particle volume fractions and reduction with respect to case without nanoparticle in percentage

φ[%]	Irreversibility [kWkg <sup>-1</sup> ]	Reduction [%]
0	60.71	0
0.5	59.64	1.762
0.7	56.11	7.577
0.8	51.85	14.594
1	49.69	18.152

 $\varphi = 0.5\%$ , while for the highest volume fraction,  $\varphi = 1\%$ , a significant decrease in total irreversibility is seen. Both the increase in the performance of the system (COP value) and the decrease in total irreversibility with the addition of the TiO<sub>2</sub> nanoadditive were obtained.

#### Conclusion

In this study, exergy analysis of a vapor-compression refrigeration system with  $TiO_2$  nanoadditive to the compressor oil was performed for R134a refrigerant. Second law of thermodynamics was used and for each of the components of refrigeration system, irreversibilities were determined. When nanoparticle was added, total irreversibility was reduced and it was significant for higher particle volume fraction. The explicit relation between the thermophysical change of the nanooil with different solid particle volume fraction and its impact on the exergy efficiency and surface erosion caused by nanooil were not considered. The cost analysis with using nanoparticles can also be considered in the future studies.

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### Nomenclature

h – enthalpy per unit mass, [Jkg <sup>-1</sup> ]	Greek symbol	
$\dot{m}$ – mass-flow rate, [kgs <sup>-1</sup> ]		
$\dot{Q}$ – heat transfer rate, [W]	$\Psi_{-}$ availability, [J]	
$\dot{S}$ – entropy, [kJ°C <sup>-1</sup> ]	Subscripts	
T – temperature, [°C] $\dot{W}$ – work, [J] $\dot{I}$ – irreversibility, [J]	in – inlet out – outlet	
· ·····, (-)	0 – reference conditions	

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